

# Mapbi3 Perovskite Solar Structure Experimental Optimization: Control of Deposition of Spinning Speed (Comparative Study of Optical and Structural Properties)

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**Abstract:** The present paper deals with the stability of halide hybrid perovskite MAPbI<sub>3</sub> (CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>) thin films by an experimental optimization by increasing the Dimethyl sulfoxide (DMSO) solvent ratio (compared to literature) in perovskite solution and the study of deposition spinning speed effect on the structural and optical properties. The samples is deposited by a spin coater in one step with 400, 1500 and 3000 rpm spinning speed. Structural and optical properties of the elaborated samples were studied using X-ray diffraction and UV-Visible spectroscopy. The main results show that compared with other works the DMSO solvent ration raise has a positive effect in the perovskite proprieties, in other hand, the best proprieties of perovskite need an optimal speed deposition; the very high or low speed has a negative effect.

**Keywords:** Perovskite solar cells, MAPbI<sub>3</sub>, spinning speed.

## 1. Introduction

The third generation of solar cells saw the appearance of perovskite cells (PSCs). The perovskite material stands as the intrinsic layer in the PIN structure of the solar cell. The halide hybrid perovskite consists of an organic-inorganic compound that adopts the general chemical formula AMX<sub>3</sub>, in which 'A' organic cation, 'M' are inorganic cation and 'X' is an anion [1]. Our perovskite MAPbI<sub>3</sub> is a semiconductor has a direct gap around 1.40 to 1.64eV [2-8], has a simple elaboration with cheap base materials [7] and possesses a high absorbance [8]. Our structure loss its organic part during or after elaboration by the light, humidity, heat or air, these factors causes a great degradation appears as a rise of inorganic element percentage (PbI<sub>2</sub>) reflecting its instability [2,9-11]. Previous improvements made by researchers to our structure was divided on three sections (impact of impurities inclusion, structure cations changing effect and the elaboration optimization) [12-21].

The improving of perovskite elaboration by reaching the optimal condition can reduce the problem of its instability, this is what we saw in our previous study (the substrate type impact) [22]. In this paper we chose the study of deposition speed effect on the structural and optical properties with an increasing of Dimethyl sulfoxide (DMSO) solvent ration (compared to literature) on the perovskite synthesis which previously unused for MAPbI<sub>3</sub> elaboration.

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## 2. Experimental Section

### 2.1. Samples elaboration

We used two powders CH<sub>3</sub>NH<sub>3</sub>I and PbI<sub>2</sub> dissolved in N, N-dimethylformamide (DMF) solvents and Dimethyl sulfoxide (DMSO) with a ration (1:4). The resulting solution was stirred at 60°C overnight. The films were prepared by spin coating on cleaned glass substrates. We used centrifugation speed 400, 1500 and 3000 rpm for 20 s. Subsequently, the freshly deposited films were annealed at 70 °C for 15 min. Regarding the XRD analysis we used a Xpert-Pro MPD of Philips diffractometer system with Cu-K $\alpha$  radiation 1.54060 Å, for UV-visible we used the Shimadzu UV-1800 UV-Vis spectrophotometer with wavelength range 300 to 700 nm, the results tracer by logiciel 'Origine6'.

### 2.2. Samples elaboration

The MAPI<sub>3</sub> gap energy was obtained from Tauc law (Eq.1) [7], where  $\alpha$  the absorption coefficient,  $h$  is the Plank constant,  $\mu$  is the frequency of light,  $A$  constant. The absorption coefficient  $\alpha$  was obtained from (Eq.2), where  $d$  the layer thickness of the thin film and  $T$  is it a transmittance. The layer thickness  $d$  was obtained from the (Eq.3) [23], where  $M$  is the mass of the thin film,  $\rho$  is MAPbI<sub>3</sub> materiel density

(4.15gr/cm<sup>3</sup>) and  $s$  is the surface of sample. The mass  $M$  obtained from (Eq.4), where  $M_i$  is the masse of substrate and  $M_f$  is the masse of sample after deposition of the film. The phases percentage were calculated from (Eq.5) where  $A$  is the PbI<sub>2</sub> integration peaks,  $B$  is CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> integration peaks and  $C$  is the diffractogram integration [24]. Preferential orientation  $R I(hkl)$  was calculated from (Eq.6) [25], where  $I_{hkl}$  present the intensity of peak. The grain size  $D$  was calculated from (Eq.7), where  $K$  is the Scherrer constant (0.9),  $\lambda$  is wave length of the X-ray beam used (1.54060 Å),  $\beta$  is the Full width at half maximum (FWHM)

of the peak and  $\theta$  is the Bragg angle. The dislocation density  $\sigma$  was calculated from (Eq.8) [26].

$$\alpha h\mu = A\sqrt{h\mu - E_g} \quad (1)$$

$$\alpha = \frac{1}{d} \ln \frac{100}{T} \quad (2)$$

$$d = \frac{M}{\rho s} \quad (3)$$

$$M = (M_f - M_i) \quad (4)$$

$$\begin{cases} P_{Pb2} (\%) = \frac{A \times 100 (\%)}{C} \\ \text{et} \end{cases} \quad (5)$$

$$\begin{cases} P_{CH_3NH_3PbI_3} (\%) = \frac{B \times 100 (\%)}{C} \end{cases} \quad (6)$$

$$R_{I(hkl)} = \frac{I_{hkl}}{\sum I_{(hkl)}} \quad (7)$$

$$D = \frac{K\lambda}{\beta \cos(\theta)} \quad (8)$$

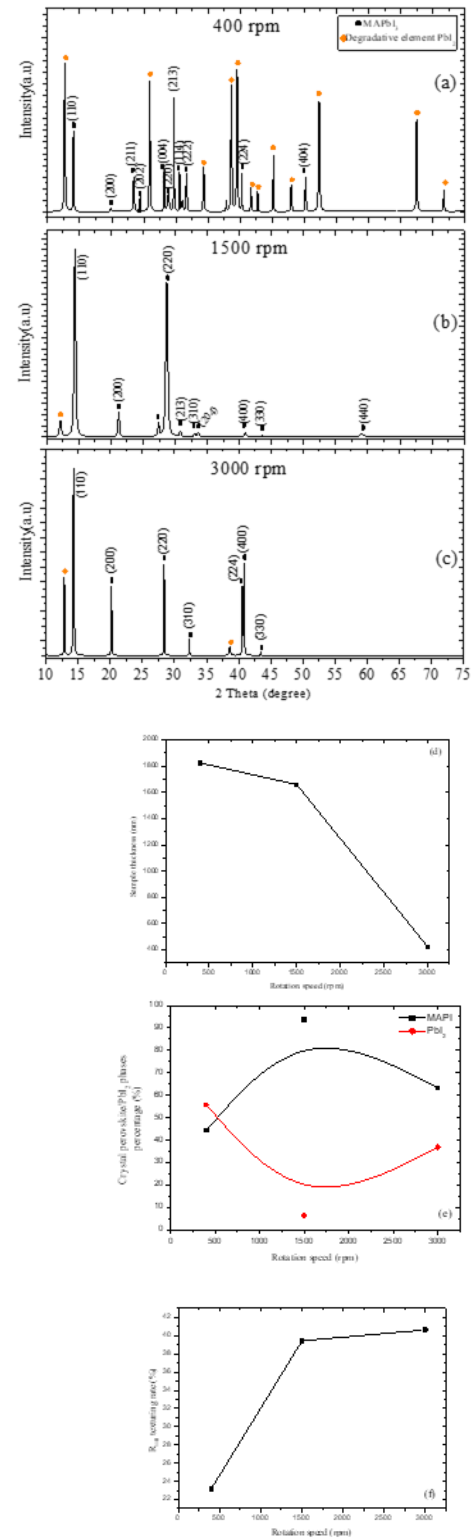
$$\sigma = \frac{1}{D^2} \quad (8)$$

### 3. Results and discussion

#### 3.1. Structural proprieties (X-ray diffraction)

Fig.1 presents the diffractogrammes of the 3 samples of MAPbI<sub>3</sub> perovskite deposited on glass substrates with 3 different deposition speeds by the spin-coating technique ((a) 3000, (b)1500 and (c) 400 rpm).

In the 3 samples, the identification of the peaks diffraction shows that the MAPbI<sub>3</sub> perovskite crystallizes according to the tetragonal phase and the line (110) exhibit as the preferred growth axis; this orientation is the most preferred for photovoltaic applications [2,13]. With 400 rpm the angles located around  $2\theta = 14.01^\circ, 19.86^\circ, 23.41^\circ, 24.38^\circ, 28.18^\circ, 28.80^\circ, 30.54^\circ, 30.96^\circ, 31.62^\circ, 40.23^\circ$  and  $50.17^\circ$ , with the deposition speed 1500 rpm the perovskite angles located around  $2\theta = 14.33^\circ, 20.30^\circ, 21.18^\circ, 23.64^\circ, 24.31^\circ, 27.53^\circ, 28.68^\circ, 30.84^\circ, 32.06^\circ, 33.51^\circ, 40.73^\circ, 43.56^\circ$  and  $59.21^\circ$  and with 3000 rpm, the perovskite angles located around  $2\theta = 14.28^\circ, 20.17^\circ, 28.56^\circ, 32.29^\circ, 40.58^\circ, 40.81^\circ$  and  $43.42^\circ$ , this results identical with Ref. [2]. All structural properties of our samples were presented in the table 1, from this table and figure 1(d, e, f), we notice that the deposition speed affect the preferential orientation percentage  $R(110)$  which increases with the spinning, the sample deposited with 3000 rpm give the highest percentage (40,65%), in other hand, the layer thickness decrease, it's may be due to the decrease in the amount of matter with the rotation speed, we notice also that the reflective element of instability (PbI<sub>2</sub>) was less with the average speed (1500 rpm) where we recorded the highest perovskite percentage (93.71%) and the lowest PbI<sub>2</sub> percentage (6.29%), the low percentage of inorganic element indicated the high level of phase purity of the perovskite [27] the sedimentation rate had no effect on other structural properties such as dislocation density and grain size but we can notice that compared to other works which used the ratio (1:1) of (DMF:DMSO), the raise of DMSO ration gave larger grains size [28].



**Fig. 1** MAPbI<sub>3</sub> structural characteristics: (a) diffractogram with 400 rpm deposition speed, (b) with 1500 rpm deposition speed, (c) with 3000 rpm deposition speed, (d) Influence of the deposition speed on the layer thickness, (e) Influence of the deposition speed on the phase percentages (MAPbI<sub>3</sub>/PbI<sub>2</sub>) in the crystal and (f) Influence of the deposition rate on the preferential orientation.

**Table 1:** Structural parameters of MAPbI<sub>3</sub> samples with different spinning speed.

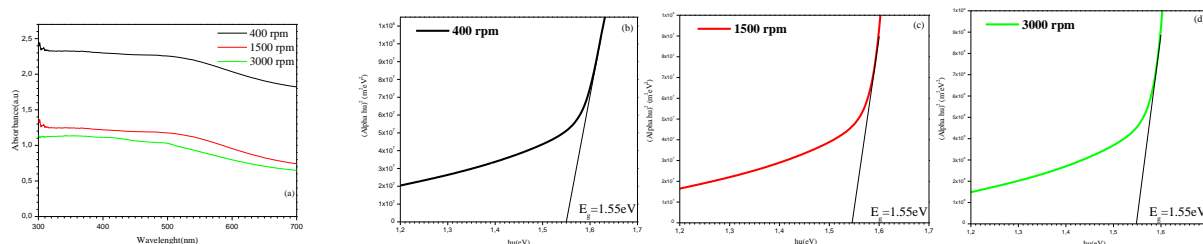
Spinning speed (rpm)	Preferred orientation R <sub>110</sub> (%)	Layer thickness d (( $\times 10^{-9}$ m)	MAPbI <sub>3</sub> percentage (%)	PbI <sub>2</sub> percentage (%)	Dislocation density $\delta$ ( $\times 10^{14}$ lines/m <sup>2</sup> )	Grain size D <sub>v</sub> (nm)
400	23,19	1807.23	44,29	55,71	0,77	113.60
1500	39,44	843.37	93.71	6.29	1,38	84.96
3000	40,65	421.69	63.23	36.77	0,44	150.53

### 3.2. Optical proprieties (UV-visible spectroscopy)

For show the change in optical properties, we present the absorbance of our samples with deposition speeds 400, 1500 and 3000 rpm in Fig2 (a), where we notice that the perovskite gives the higher absorbance in the sample deposited with the lowest spinning speed (400 rpm), then, the absorbance begins to decrease with increasing deposition speed in all wavelength range. These results may be due to the amount of material which loss with the rise of rotation speed where the spin coating method considerably changes the density of the material (the solution remaining during

the deposition process) [29].

Fig.2 (b, c, d) presents the variation of  $(\alpha h\nu)^2$  versus photon energy  $(h\nu)$  (Tauc plots) of samples deposited with 400,1500 and 3000 rpm respectively. The intersection point with the  $(h\nu)$  axis of the linear extrapolation of this curve gives the band gap energy. We notice that the gap energy was not affected by changing the deposition speed. The band gap is determined to be equal to 1.55 eV for all samples with the different speed's deposition. It is noted that these values are in close agreement with the perovskite gap [11].

**Fig. 2** MAPbI<sub>3</sub> optical characteristics: (a) Samples absorbance, (b) Tauc diagram with 400 rpm deposition speed, (c) Tauc diagram with 1500 rpm deposition speed and (d) Tauc diagram with 3000 rpm deposition speed.

## 4. Conclusion

For a maximum functional diversity in the perovskite MAPbI<sub>3</sub> and stable structure we need an optimal deposition speed. The perovskite characteristics are negatively affected such as (low absorbance and the insufficient thickness) if the deposition speed is too high and the very low deposition speed gives high percentage of inorganic element PbI<sub>2</sub> (instable structure) and large thickness resulting high dislocation density.

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